

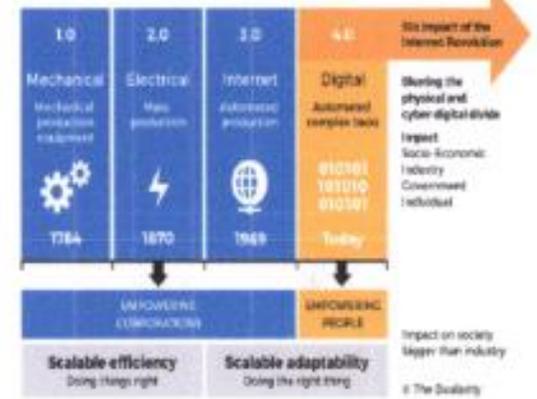
Tren Bioteknologi Era Revolusi Industri 4.0: Teknologi Enzimatik

Ditulis oleh:
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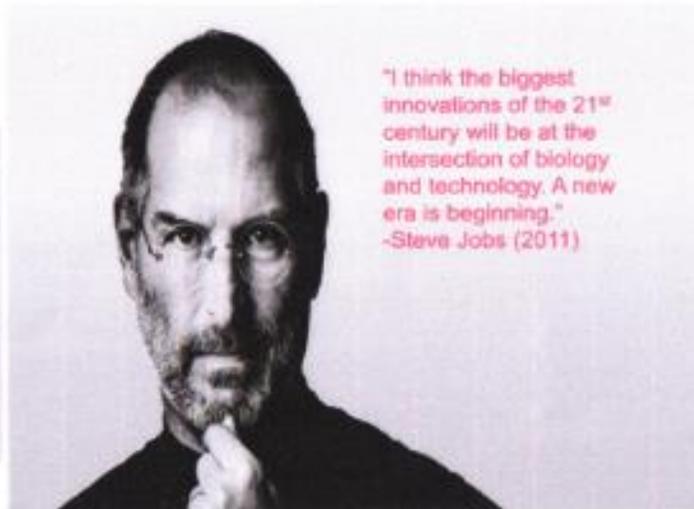
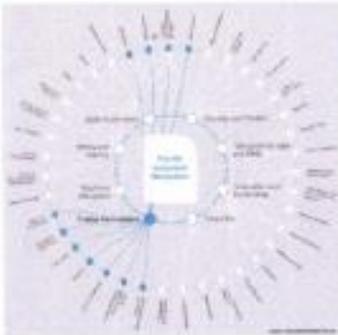
- **Biomaterial and Enzyme Technology Research Group**
 - > 50 peer reviewed publications
 - 2 book chapters
- **Antioxidant measuring system**
- **Polymer processing**
 - Functionalisation of synthetic and natural polymers
 - Enzymes: Hydrolases, Oxidoreductases
- **Enzymology**
 - New enzymes, screening procedures
 - Biochemical characterisation
 - Microtransformation pathways
- **Contact: endry@bio.its.ac.id**

Welcome to the Industrial Revolution 4.0



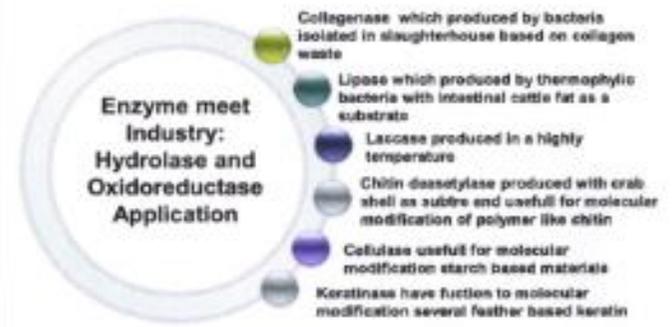
INDUSTRY 4.0 FUSING DIGITAL, PHYSICAL, & BIOLOGICAL

"The fourth industrial revolution, is not only about the smart and connected machines and systems. Its scope is much wider. Occurring simultaneously are waves of further breakthroughs in areas ranging from gene sequencing to nanotechnology, from renewables to quantum computing. It is the fusion of these technologies and their interaction across the physical, digital and biological domains that make the fourth industrial revolution fundamentally different from previous revolutions." Klaus Schwab, founder and Executive Chairman of the World Economic Forum

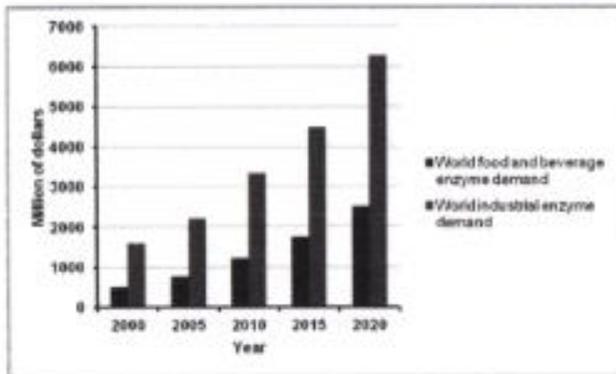


ENZYME. NOWDAYS

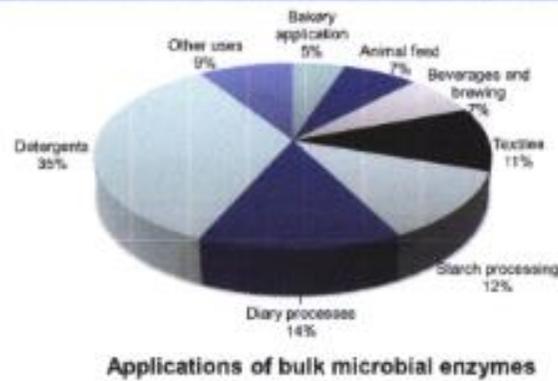
Biomaterial and enzyme technology research group well managed to develop several enzymes



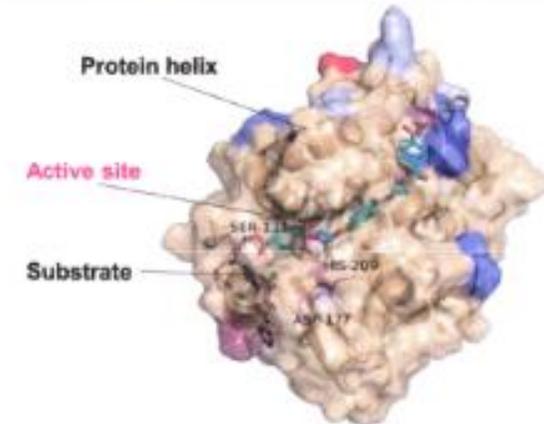
Kebutuhan Enzim Dunia



APLIKASI ENZIM



Apa sebenarnya enzim itu?

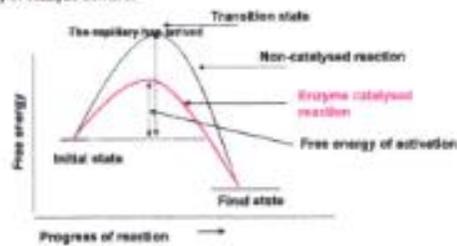


Why Enzymes?

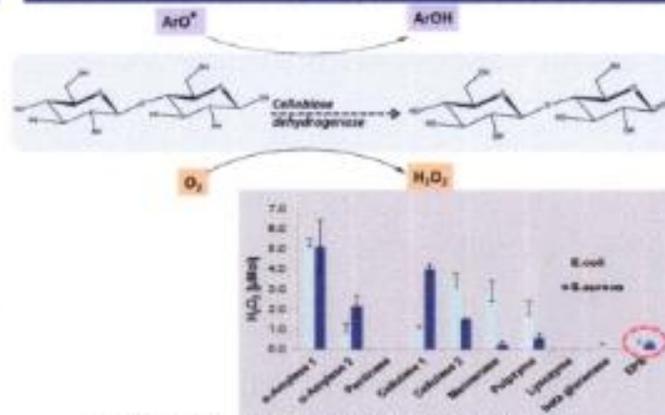
Enzymes are highly specific biocatalysts
Only perform certain transformations while leaving other substrates intact

Avoids use of hazardous chemicals
Enzymes work under "mild" conditions

Enzymes are fully biodegradable
since they are proteins
since they are used only in catalytic amounts



Enzymes as disinfectants



Wang et al. *Enzymes and Microbial Technology* <https://doi.org/10.1016/j.enzmictec.2017.11.007>
Thalinger et al. *Biotechnology Journal* <https://doi.org/10.1002/biot.7930013>

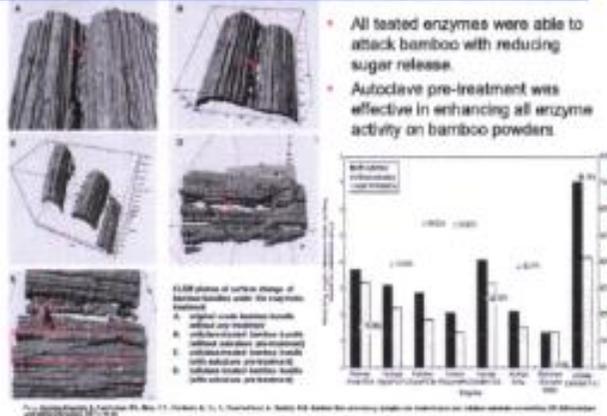
Environmental friendly processes

Deezing Acids, alcohols or oxidizing agents Amylases
Scouring Sodium hydroxide, hydrogen peroxide, Sodium hypochlorite, T > 90°C, pH > 10 Pectinase, Cellulase, Pectinase, Laccase, Cellulase, Ta 70°C, pH > 9
Bleaching Chlorine bleaches, hydrogen peroxide, Glucose oxidase, Lignin peroxidase, Manganese peroxidase, Laccase
Stone washing 1 Kg of stones for 1 Kg of jeans Cellulase and laccase-mediator systems
Bleach Clean-up Catalase
Waste effluents, dyes oxidoreductases

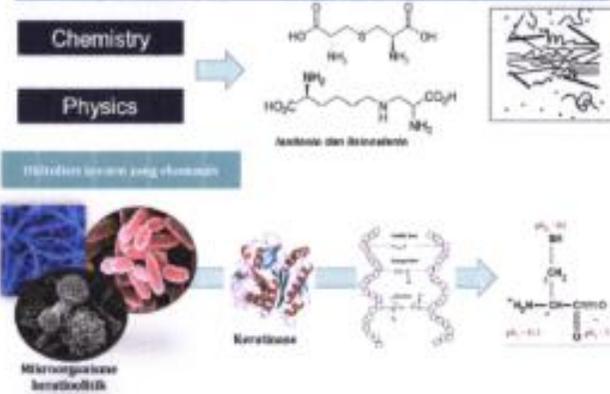


Wang et al. <https://doi.org/10.1016/j.enzmictec.2017.11.007>
Fu et al. 2012, *Microbial and Molecular Enzymes* 10: 27-47
Harrington et al. *Biotech Biochem* 84 (1992) 266-269
Harrington et al. 2002 *Water Res* 36 (1246)-1248

Bamboo fibre processing



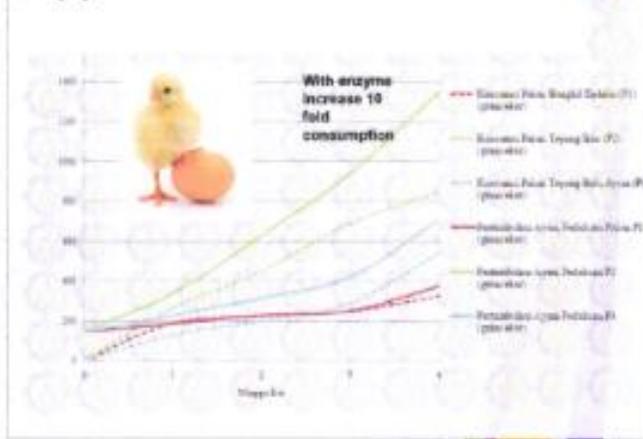
Keratinase for increasing of soluble protein



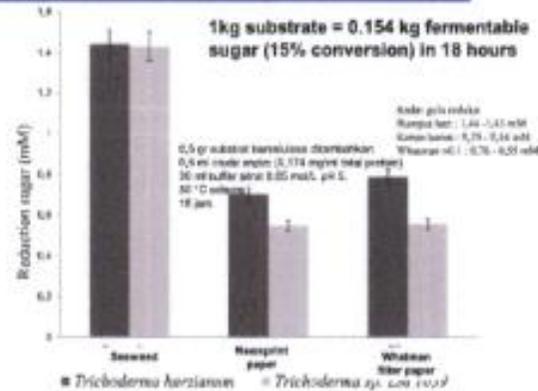
Biocatalysis keratin based waste



Application in chicken feed



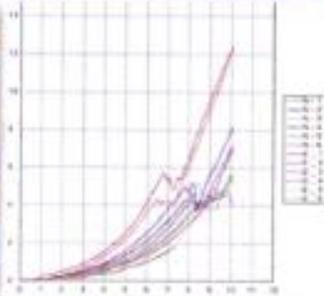
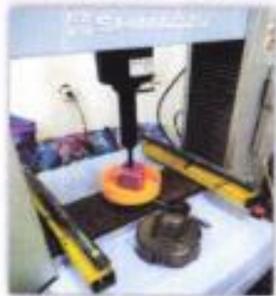
Cellulase on various paper based substrate



Collagenase for meat tenderizing



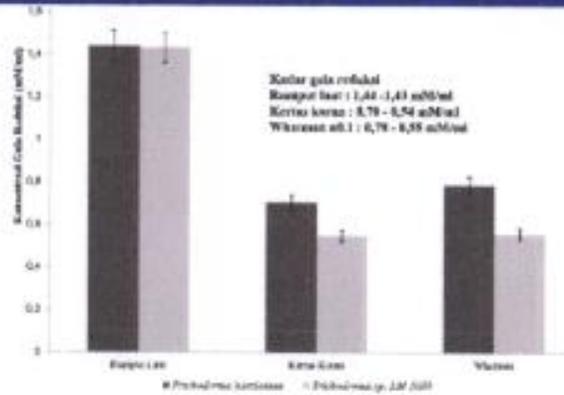
Meat softness test



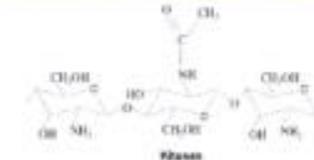
21% tenderized meat in 15 min by collagenase treatment at room temperature



Aplikasi Enzim Pada Substrat



Enzymatic production of chitosan



Chemical Process



Biological Process
Chitin Deacetylase
EC 3.5.1.41



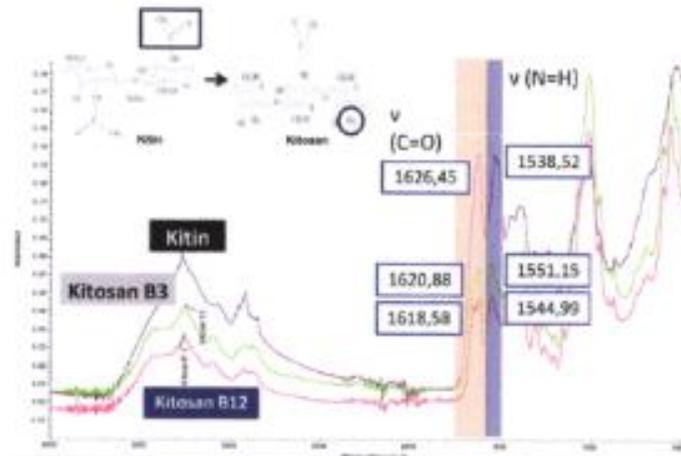
Degradasi Polimer
ACAK

Degree of deacetylation by different source of chitin deacetylase

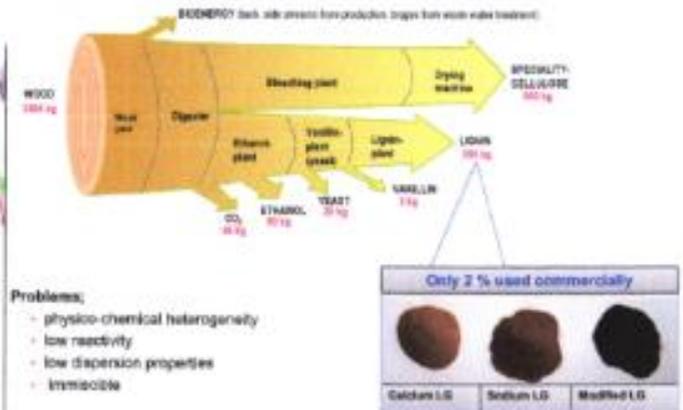
Analysis	Degree of deacetylation after enzymatic treatment				
	with standard	Kitosan B1	Kitosan B2	Kitosan B3	Kitosan B4
Glucosamine standard (µM)	219	585	454	483	333
Degree of deacetylation (%)	< 30	63	52	55	59

Synowiecki et al., (2003), degree of deacetylation of chitin more than 50% has transformed become chitosan

FTIR spectra of enzymatic chitosan



Lignin



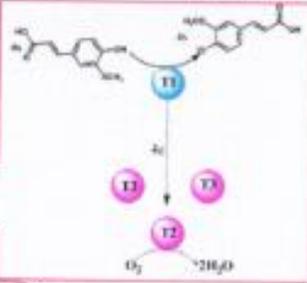
Laccase

Laccases catalyze the oxidation of various aromatic compounds while simultaneously reducing O_2 to H_2O

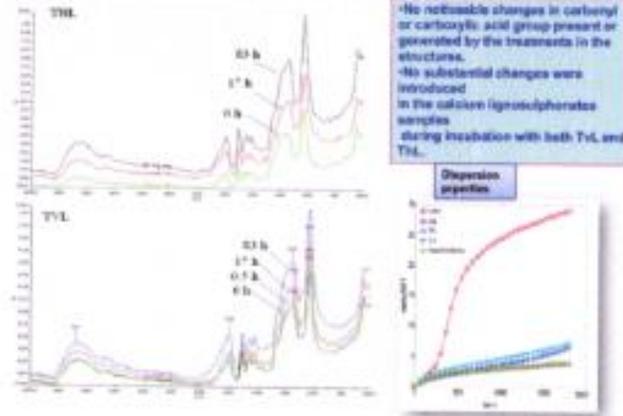
Distribution: - Fungi, plants, insects, bacteria



Pierick et al. 2002, JBC 277, 37665-37669
Collaboration partner in BIORENEW project

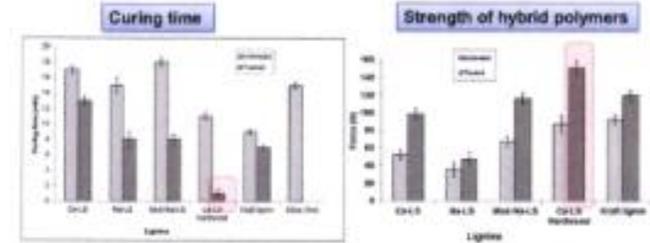


FTIR and chemical analysis



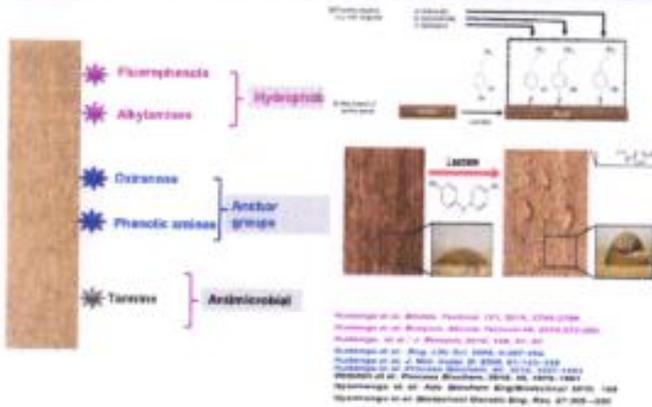
Synthesis of siloxane-lignin hybrid functional polymers

Lignin + laccase + HBT + TEOS

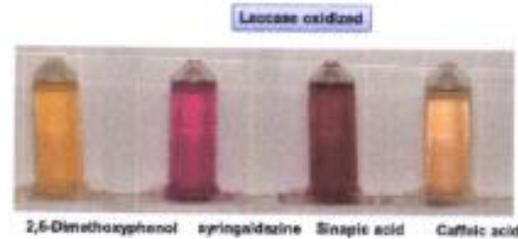


Nugroho Prasetyo et al 2022, *Biotechnology Journal* 7(2):284-82

Other activity targeting lignin in lignocellulose polymers

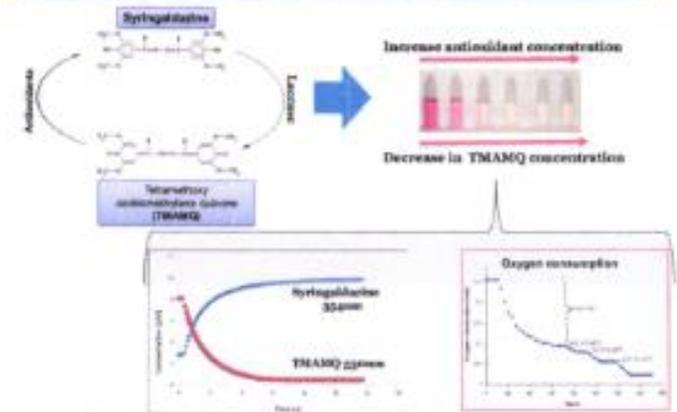


Monitoring human exposure to Environmental pollutants



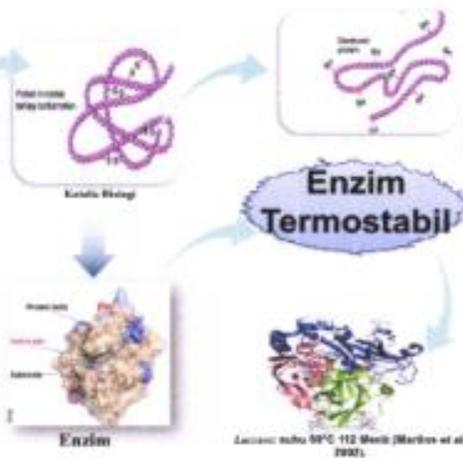
Baron et al. *Analytical and Bioanalytical Chemistry* 2005, 292, 2766-2768
Nugroho Prasetyo et al. *Biotechnology Journal* 2022, 16, 284-292
Nugroho Prasetyo et al. *Journal of Applied Microbiology* 2022, 133, 1221-1230
Nugroho Prasetyo et al. *Journal of Applied Microbiology* 2022, 133, 1221-1230
Nugroho Prasetyo et al. *Journal of Applied Microbiology* 2022, 133, 1221-1230
Nugroho Prasetyo et al. *Journal of Applied Microbiology* 2022, 133, 1221-1230
Nugroho Prasetyo et al. *Journal of Applied Microbiology* 2022, 133, 1221-1230

Measuring antioxidant capacity using TMAMQ

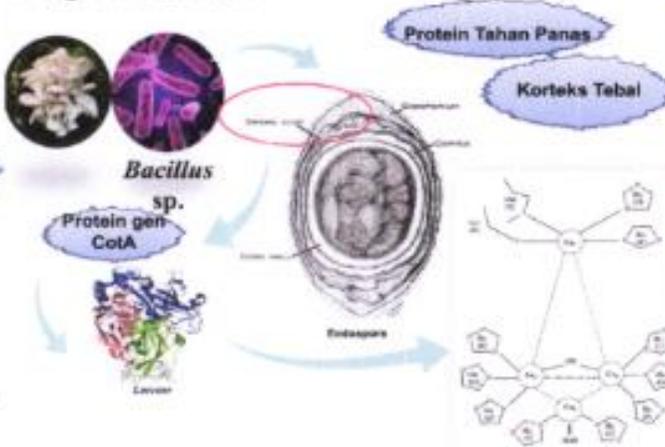


Nugroho Prasetyo et al 2009, *Analytical & Bioanalytical Chemistry*

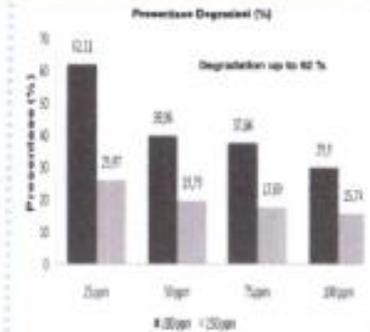
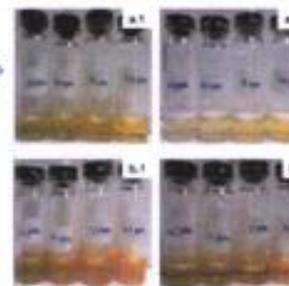
Katalis



Penghasil Laccase



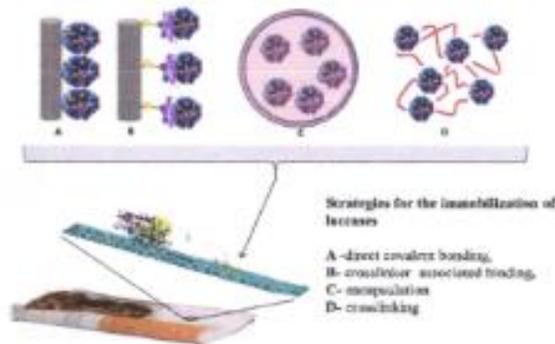
Aplikasi Biobleaching Pewarna Azo Orange II



a1 Laccase Komersial Tebal 100 ppm
a2 Laccase Komersial Tebal 200 ppm
b1 Laccase 117 Substrum Terdegradasi 0h
b2 Laccase 117 Substrum Terdegradasi 24h

Gambar 4.6 Presentase Degradasi Zat Warna Azo Orange II dengan Variasi Konsentrasi

Laccase immobilization strategies onto cellulose acetate



Strategies for the immobilization of laccases

- A- direct covalent bonding,
- B- crosslinker associated binding,
- C- encapsulation
- D- crosslinking

Cigarettes smoke toxicant removal

Different strategies used to remove cigarette smoke toxicants	
Biocatalysis	Enzymatic working
Acrylamide	Copolymerization with hydroquinone
β-Aminopropylamine	Direct oxidation
(Soybean/leek)	
Catechol	Direct oxidation
Hydroquinone	Direct oxidation
Oxetanedehyde	copolymerization
Water acids	Copolymerization with hydroquinone, catechol and methylglyoxal
BNB	Copolymerization with hydroquinone, catechol and methylglyoxal
BNB	Copolymerization with hydroquinone
Acetaldehyde	Copolymerization with hydroquinone, catechol and methylglyoxal
Acetone	Copolymerization with hydroquinone, catechol and methylglyoxal
Benzene	No oxidation and not possible to copolymerize
Benzaldehyde	Copolymerization with hydroquinone, catechol and methylglyoxal
Formaldehyde	Copolymerization with hydroquinone, catechol and methylglyoxal

Comparative Biocatalysis to the presence of laccase phenolic substrates						
Sample/Method	Formaldehyde concentration with laccase (µM)	Wt. Den.	Successive %	Formaldehyde concentration without laccase (µM)	Wt. Den.	Decrease in %
without NaOH/without any agent	93.57	0.45	50.00	93.57	0.45	1.00
Hydroquinone	93.27	2.12	55.81	94.47	0.47	1.91
Methylhydroquinone	93.81	0.81	2.01	94.26	4.76	92.38
Trihydroxyhydroquinone	94.49	3.63	16.29	94.35	1.11	2.32
Catechol	93.42	1.83	1.41	94.32	3.38	11.65
Methylcatechol	93.75	1.33	1.91	92.14	0.65	10.07
2,4-Dihydroxyacetone	112.59	0.54	15.11	95.11	1.02	43.55

Source: Choudhry, S., Sarkar, S., Laha, S., Choudhary, V., Kumbhkar, S.S., Das, G.R., L. (2016) Removal of formaldehyde from cigarette smoke by using laccase immobilized on cellulose acetate. *Chemosphere*, 151: 104-110.

Future perspectives

- Cold resistant collagenase
- Novel natural mediator for laccase
- Biobleaching on natural cotton
- Development of enzyme based analytical method for antiaging agent
- Enzymatic based fertilizer
- Enzyme based gelling agent
- Enzyme base grouting materials
- Enzyme based phenolic sensor
- Enzymatic modification of drug carrier

Partners



3 postdoc
2 PhD
3 Master
25 Bachelor



Thank you